

# Au°-Nanoparticles: Control Size and Morphology Stabilized by Tripodal Phosphine Based Ligands and Their Antimicrobial Activity

Bibek Jyoti Borah<sup>1</sup>, Archana Yadav<sup>2</sup>, and Dipak Kumar Dutta<sup>1</sup>

<sup>1</sup> Materials Science and <sup>2</sup> Biotechnology Divisions, North East Institute of Science and Technology (CSIR), Jorhat 785006, India

Metal nanoparticles have attracted great attention due to their potential applications in the fields like electronics, opto-electronics, bioscience, catalysis, etc. New ligands 1,1,1-tris (diphenylphosphinomethyl) ethane  $[CH_3C(CH_2PPh_2)_3]$  1 and 1,1,1-tris-(diphenylphosphino methyl) ethane trisulphide  $[CH_3C(CH_2P(S)Ph_2)_3]$  2 for stabilization of Au $^\circ$ -nanoparticles having small core diameter were prepared. The Au $^\circ$ -nanoparticles are more active against Gram(+) bacteria, e.g., Staphylococcus aureus than Gram(-) bacteria. Among the two Au $^\circ$ -nanoparticles, the ligand 1 stabilized particles shows higher activity.

**Keywords:** Au°-Nanoparticles, Antimicrobial Activity, Phosphine, Surface Plasmon, Single-Crystalline.

## 1. INTRODUCTION

Metal nanoparticles have attracted great attention due to their potential applications in the fields like electronics, opto-electronics, bioscience, catalysis, etc.<sup>1-5</sup> Ligands stabilized metal nanoparticles are considered as important building blocks. A numbers of phosphines and thiols stabilized metal nanoparticles were studied<sup>6</sup> but metal nanoparticles particularly Au°-nanoparticles stabilized by tripodal phosphines and its chalcogen functionalized have not been reported so far. We report new ligands 1,1,1-tris (diphenylphosphinomethyl) ethane [CH<sub>3</sub>C(CH<sub>2</sub>PPh<sub>2</sub>)<sub>3</sub>] 1 and 1,1,1-tris-(diphenylphosphinomethyl) ethane trisulphide [CH<sub>3</sub>C(CH<sub>2</sub>P(S)Ph<sub>2</sub>)<sub>3</sub>] 2 for stabilization of Au°-nanoparticles having small core diameter and their antimicrobial activity against some Gram(+) and Gram(-) bacteria.

## 2. MATERIALS AND METHODS

HAuCl<sub>4</sub> · 3H<sub>2</sub>O, 1,1,1-tris (diphenylphosphino-methyl) ethane, elemental sulfur, Tetra-n-octylammoniumbromide and NaBH<sub>4</sub> were purchased from M/S Sigma Aldrich, USA. Phosphine stabilized Au°-nanoparticles were synthesized following the procedure of Hutchison et al.<sup>5</sup> The

synthesized Au°-nanoparticles were characterized by UV-Visible spectroscopy, XRD, TEM and HRTEM, etc. techniques and evaluated for antimicrobial activity by standard Agar-well-diffusion method.

### 3. RESULTS AND DISCUSSION

The broad surface plasmon bands (Fig. 1(A)) of Au°-ligands (1 and 2) at around  $\sim$ 520 nm indicate that the particles stabilized by ligand 1 are smaller than that of Au°-ligand 2.<sup>1</sup>

The TEM images (not shown) for ligands 1 and 2 stabilized Au $^{\circ}$ -nanoparticles indicate narrow size distribution of  $1.6 \pm 0.4$  nm and  $3.0 \pm 0.7$  nm respectively. The HRTEM pattern (not shown) of Au $^{\circ}$ -nanoparticles stabilized by ligand 1 shows their single-crystalline nature.

The XRD peaks of Au°-nanoparticles stabilized by ligand 1 could not be detected due to the small size (<2 nm), while ligand 2 stabilized nanoparticles shows diffraction peaks (Fig. 1(B)) corresponding to the (111), (200), (220) and (311) surfaces of fcc gold, indicate the products are composed of crystalline gold.

The Au°-nanoparticles are more active against Gram(+) bacteria, e.g., *Staphylococcus aureus* than Gram(-) bacteria. Among the two Au°-nanoparticles, the ligand 1 stabilized one shows higher activity (Table I).

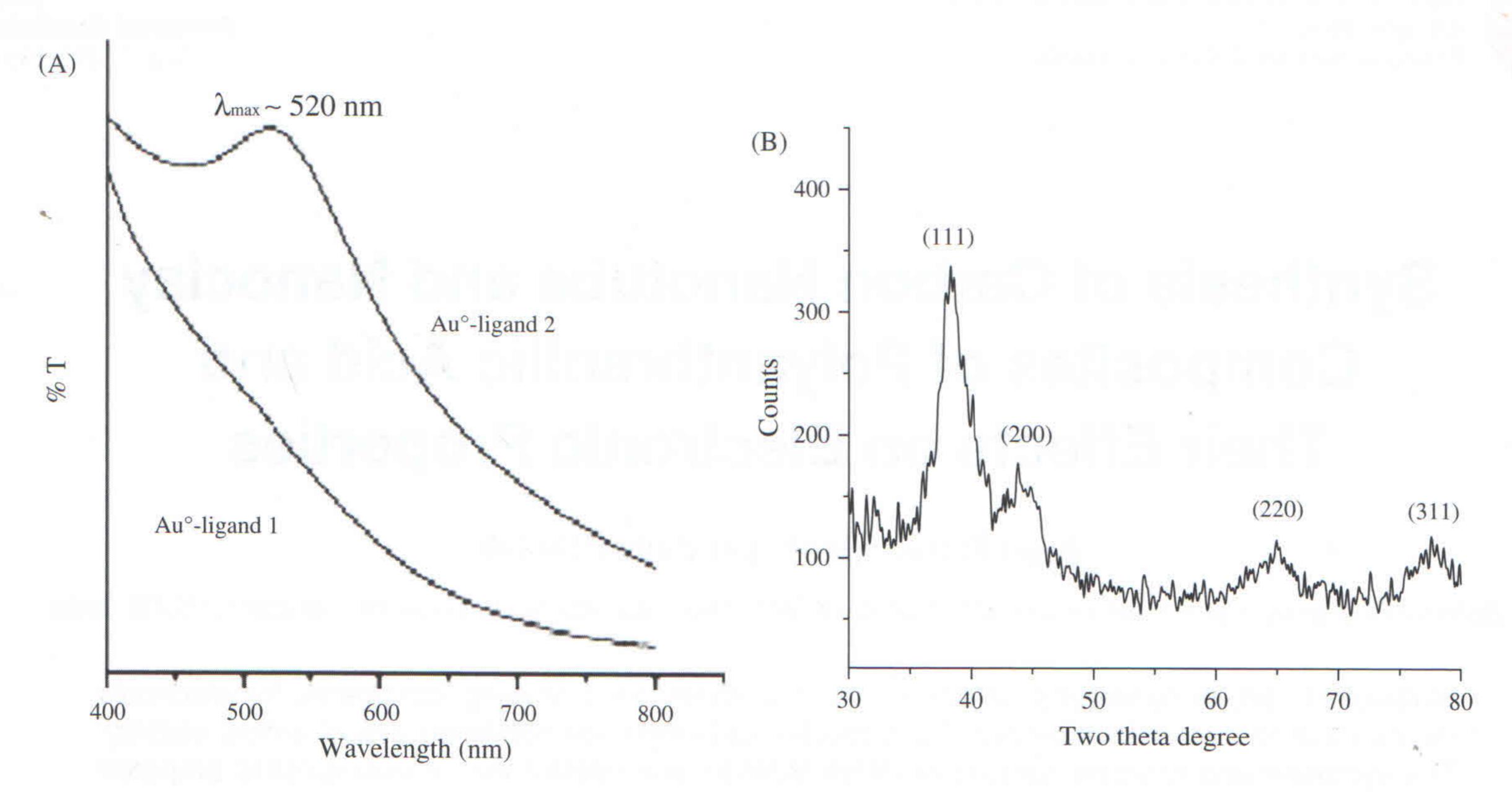


Fig. 1. (A) UV-Visible spectra and (B) Powder XRD pattern Au°-nanoparticles stabilized by ligand 2.

Table I. Antimicrobial activities of Au°-nanoparticles.

Bacteria used	Inhibition zone diameter (mm)			
	Au° ligand 1	Au° ligand 2	Ligand 1	Ligand 2
Staphylococcus	17	11	0	0
Escherichia coli	10	8	0	0

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### References and Notes

1. C. N. R. Rao, G. U. Kulkarni, P. J. Thomas, and P. P. Edwards, Metal nanoparticles and their assemblies. *Chem. Soc. Rev.* 29, 27 (2000).

- O. S. Ahmed and D. K. Dutta, Generation of metal nanoparticles on montmorillonite K10 and their characterization. *Langmuir* 19, 5540 (2003).
- 3. O. S. Ahmed and D. K. Dutta, *In situ* generation of metal clusters in interlamellar spacing of montmorillonite clay and their thermal behavior. *Thermochim. Acta* 395, 209 (2003).
- **4.** B. J. Borah, D. Dutta, and D. K. Dutta, Controlled nanopore formation and stabilization of gold nanocrystals in acid-activated montmorillonite. *Appl. Clay Sci.* 49, 317 (**2010**).
- 5. W. W. Weare, S. M. Reed, M. G. Warner, and J. E. Hutchison, Improved synthesis of small ( $d_{\rm core} \sim 1.5$  nm) phosphine stabilized gold nanoprticles. J. Am. Chem. Soc. 122, 12890 (2000).
- 6. M. Brust, M. Walker, D. Bethell, J. D. Schiffrin, and R. Whyman, Synthesis of thiol-derivatised gold nanoparticles in a two-phase liquid-liquid system. *J. Chem. Soc. Chem. Commun.* 7, 801 (1994).
- 7. J. Sun, M. Guan, T. Shang, C. Gao, Z. Xu, and J. Zhu, Selective synthesis of gold cuboid and decaherdal nanoparticles regulated and controlled by Cu<sup>2+</sup> ions crystal. *Growth & Design* 8, 906 (2008).

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